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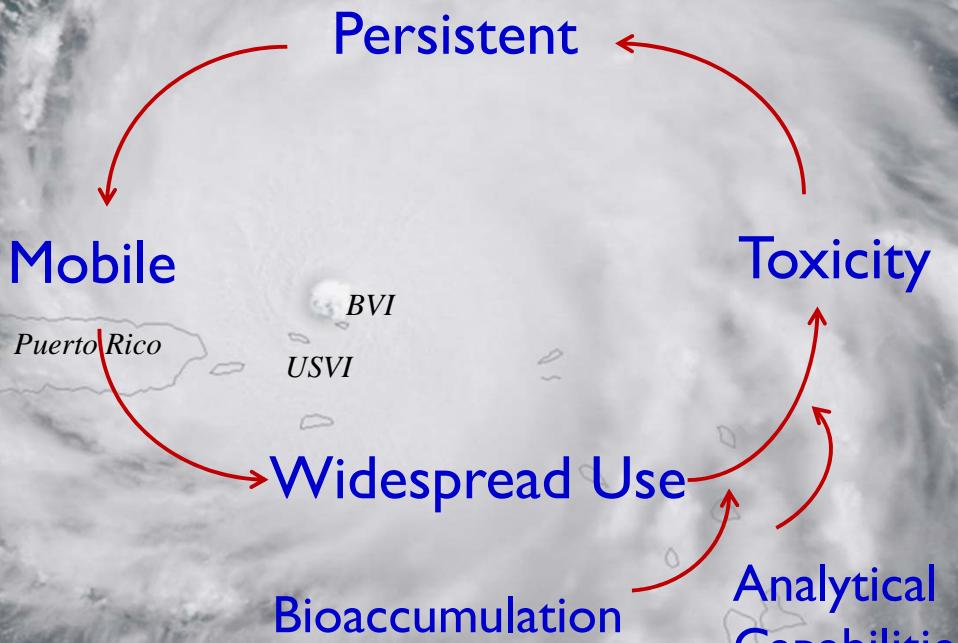
**EMERGING CONTAMINANTS  
S U M M I T**

# In Situ Sequestration of Per- and Polyfluoroalkyl Substances (PFAS) from Contaminated Groundwater

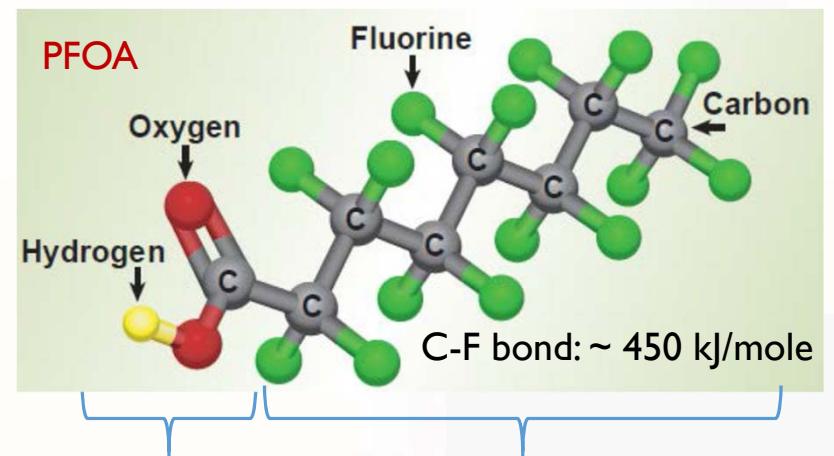
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Natalie Capiro (Tufts), Bill Arnold (UMinn), Jim Hatton (Jacobs), Matt Simcik (UMinn)*

## PFAS: A Perfect Storm (Irma)



*Conventional in situ technologies  
are not well suited for PFAS:*



Hydrophilic “head” group      Hydrophobic/lipophilic “tail”

- Low Volatility (Thermal)
- Recalcitrant (Bioremediation, ISCO)
- Surfactant (Interfaces/Sorption?)

# Potential PFAS Treatment Options

Technology	Summary of PFAS Treatment		
	PFOA	PFOS	Application
Chemical Oxidation	YES <sup>1,2,3,4</sup>	Partial <sup>4</sup>	Ex-Situ (Reactor) In-Situ (?)
Chemical Reduction	YES <sup>5,6</sup>	Partial <sup>5,6</sup>	Ex-Situ (Reactor) In-Situ (?)
Electrochemical	YES <sup>7,8</sup>	YES <sup>7</sup>	Ex-Situ (Reactor) In-Situ (?)
Sorption/Sequestration	YES <sup>9,10</sup>	YES <sup>9,10</sup>	Ex-Situ (GAC/Resin) <del>In-Situ (Injection/Barrier Wall)</del>
Biological Treatment	NO	NO	Unlikely, PFAS may inhibit chlorinated solvent degradation

<sup>1</sup>Liu et al. 2012, Sep and PurfTech: Heat-activated persulfate

<sup>2</sup>Mitchell et al 2013, ES&T Letters: Catalyzed hydrogen peroxide

<sup>3</sup>Vecitis et al. 2009, *Front. Environ. Sci. Engin. China*: UV with TiO<sub>2</sub>

<sup>4</sup>Park et al., 2016, *Chemosphere*: Heat-activated persulfate

<sup>5</sup>Ochoa-Herrera et al. 2008, ES&T: Ti(III)-citrate and Vit B<sub>12</sub>

<sup>6</sup>Wang et al., 2017, *Chem. Eng. J.*, Photocatalytic reactivity

<sup>7</sup>Schaefer et al., 2015, *J. Haz. Mater.*: TiRuO<sub>2</sub> anode

<sup>8</sup>Zhou et al., 2017, *J. Electro. Chem.*: PbO<sub>2</sub> electrode + PVDF

<sup>9</sup>Zhang et al., 2016, *Chemosphere*: GAC

<sup>10</sup>Yu et al., 2009, *Water Res.*: GAC and resin



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# Conventional Pump & Treat PFAS Remediation

Washington County, MN



3M Settles Minnesota Lawsuit for \$850M (Feb 20, 2018)

#ECSUM18



MN Department of Health/3M

Breakthrough times:

- PFBA = 30 days
- PFOA = 286 days
- PFOS = 550 days



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# Commercially Available (Proprietary) PFAS Sorbents



## RemBind™-Tersus

Activated carbon, aluminum hydroxide, organic matter and other additives, intended for near surface soil mixing



## PlumeStop® Liquid Activated Carbon™—Regenesis

Activated carbon (1-2µm) suspended in water dispersed with organic polymer

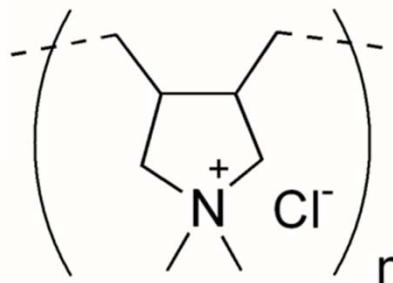
- ❖ Limited independent verification
- ❖ Limited data (e.g., mass balance)
- ❖ In situ delivery issues rarely addressed



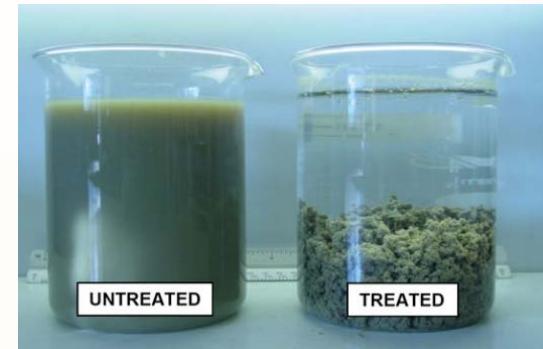
# *Coagulant polymers (cationic surfactants)*

SERDP Project ER-2425 (Simcik, Arnold, Pennell, Hatton)

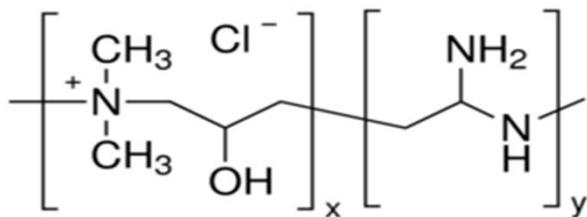
## Poly-DADMAC (PDM)



- Accepta 4351
- ~ 28% OC
- Quaternary Amine
- diallyl dimethylamine
- MW ~ 350,000



## Polyamine (PA)

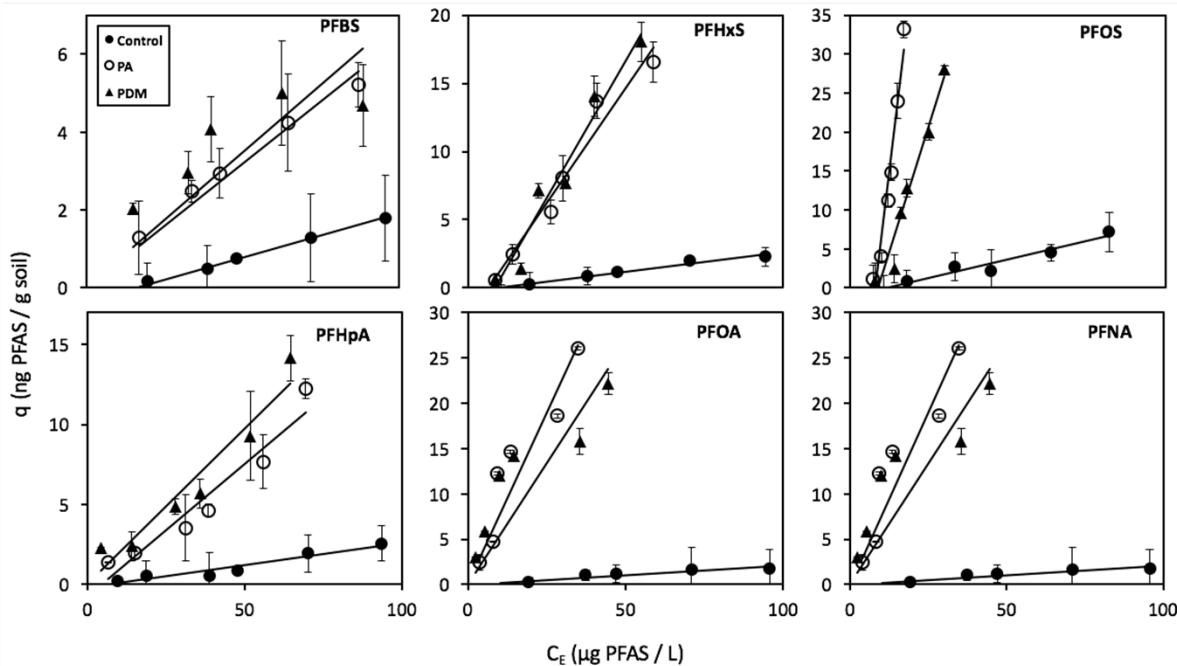


- Accepta 4350
- ~ 26% OC
- Quaternary Amine
- epichlorohydrine and dimethylamine
- MW ~ 240,000



# Batch PFAS Sorption Tests: PDM and PA

40-50 mesh Ottawa Sand

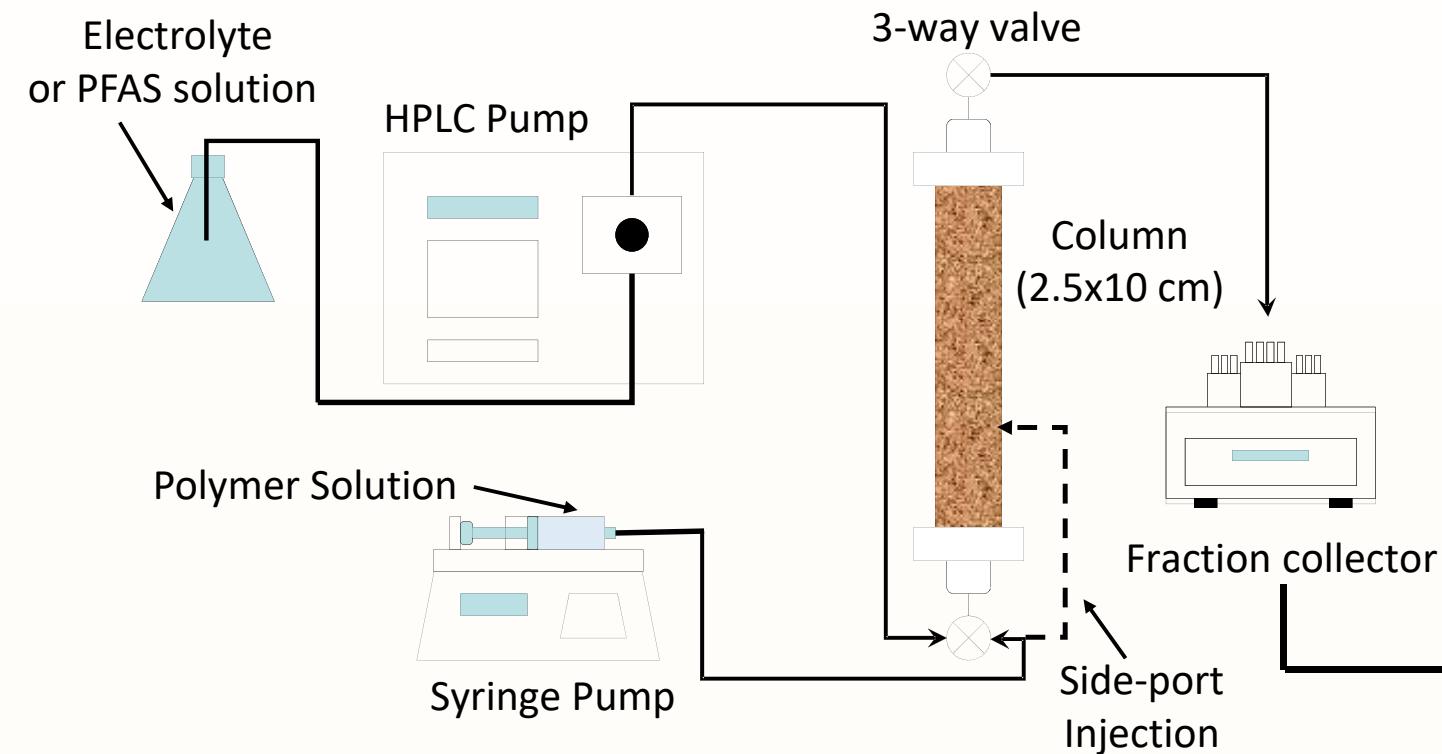


- ❖ Addition of PDM or PA increased PFAS sorption by Ottawa sand and Tinker AFB soil
- ❖ PFAS Sorption coefficients were 3.8 to 45-fold greater in the presence PDM or PA
- ❖ PDM yielded slightly greater sorption improvements than PA

Aly et al., 2018, J. Env. Eng., (in press)



# Schematic Diagram of 1-D Column System



- (1) Non-reactive tracer test (pulse injection), (2) Inject PDM+PAC suspension,
- (3) Inject background electrolyte, (4) Inject PFAS solution (e.g., 100 ug/L PFOS)

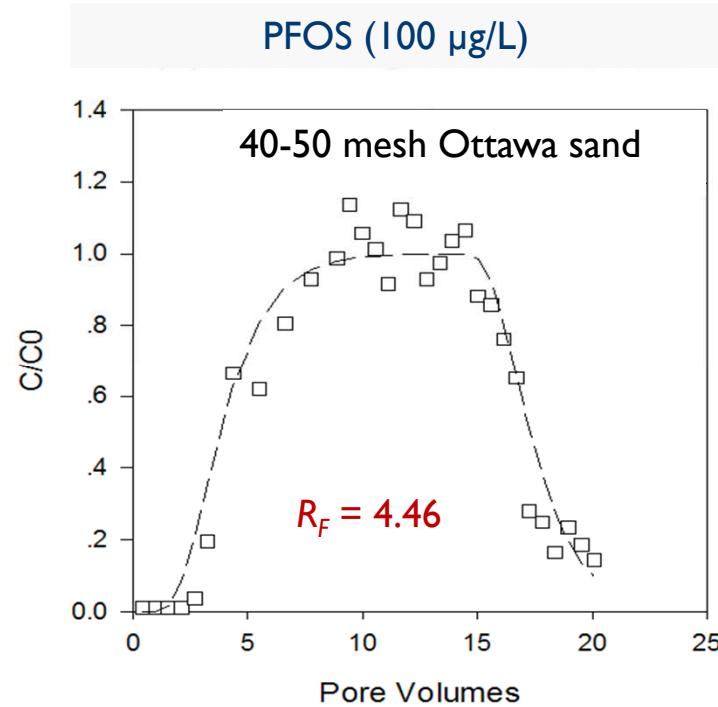
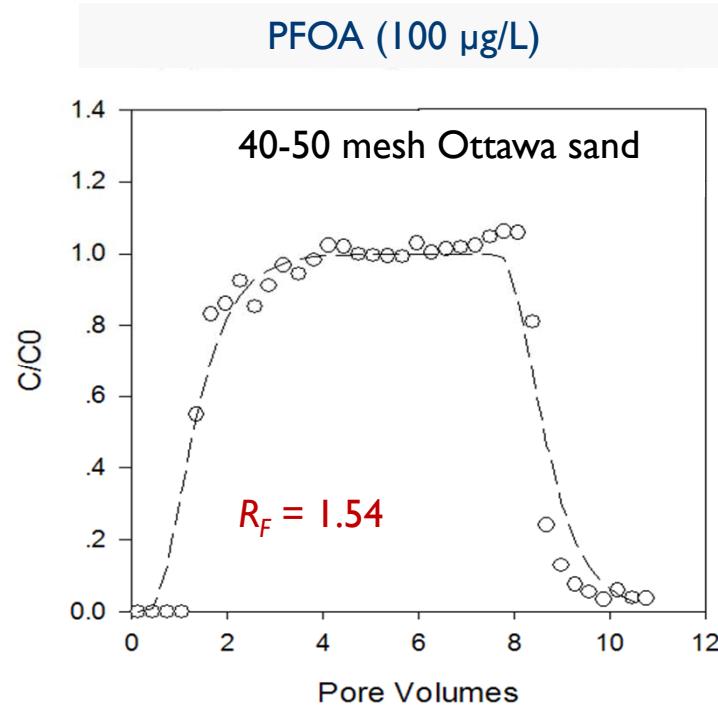
Waters Xevo TQ-S



Effluent samples  
analyzed by LC-MS/MS



# PFAS Column Tests: Control (w/o PDM or PA)

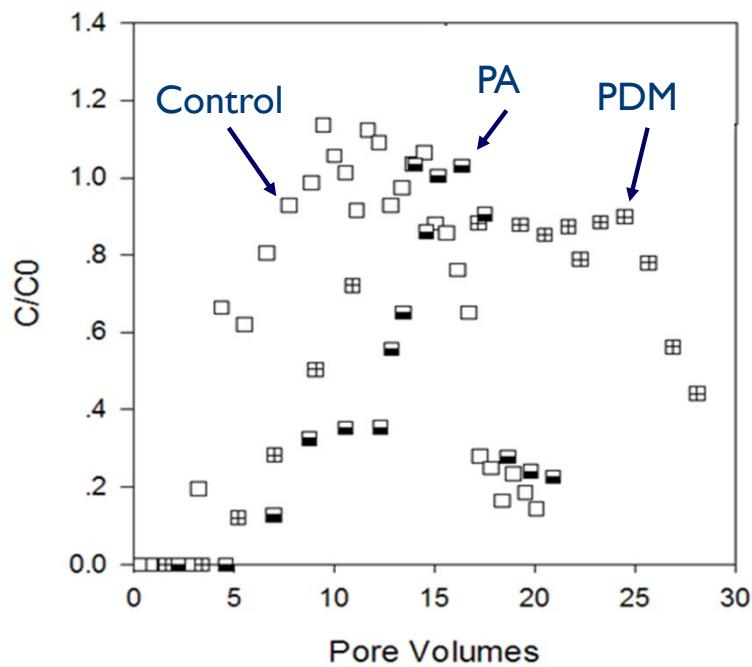


Aly et al., 2018, J. Env. Eng., (in press)

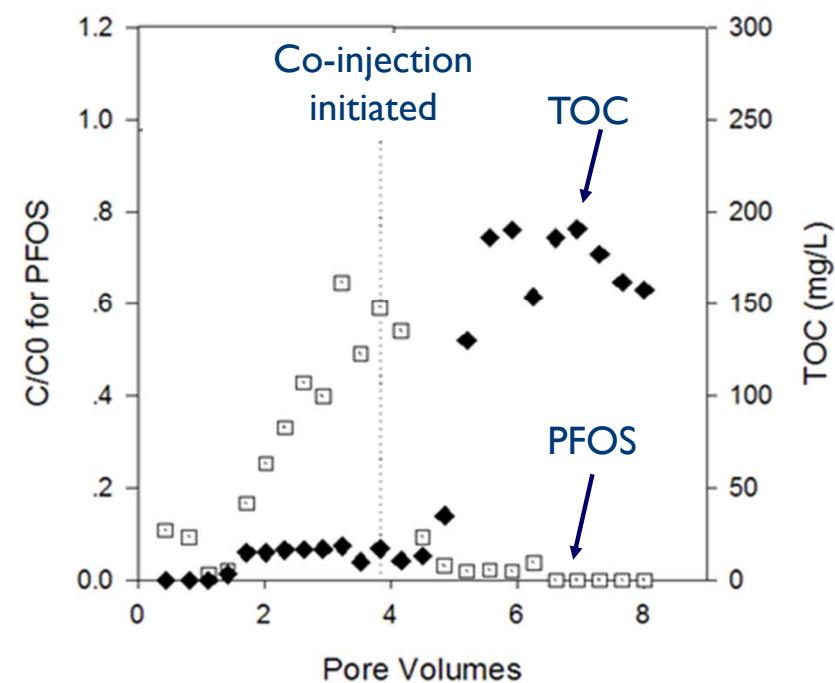


# Pretreated and Side-Port Co-injection: PDM + PFOS

Pretreated 40-50 mesh Ottawa sand (100 ug/L PFOS)



Side-Port Co-injection 40-50 mesh Ottawa sand (100 ug/L PFOS)



Aly et al., 2018, J. Env. Eng., (in press)

## Summary of PFAS Column Results: PDM or PA

	PFAS Retention (ng/g sand)	PFAS Retention (%)	Enhancer retention (ng/g sand)	Retained PFAS/Retained Enhancer mass ratio
<b>PFOA</b>				
control	5.97	8.95%	n/a	n/a
PDM pre-treatment	8.38	9.50%	21.68	0.11
PA pre-treatment	20.10	21.16%	50.83	0.28
PDM side port injection	30.62	49.22%	60.3	0.41
<b>PFOS</b>				
control	4.20	6.23%	n/a	n/a
PDM pre-treatment	91.94	34.91%	21.68	4.04
PA pre-treatment	43.59	50.04%	77.00	0.51
PDM side port injection	196.06	83.22%	42.86	4.47

## To improve performance....combine Powdered Activated Carbon (PAC) with polyDADMAC (PDM)

- ❖ PDM acts to stabilize PAC in suspension, facilitates delivery
- ❖ Both PDM and PAC can serve as sorbents (wide range of effectiveness)

1 g/L PAC

1 g/L PAC + 5 g/L PDM

1 g/L PAC

1 g/L PAC + 5 g/L PDM



DARCO® 100 mesh (150 µm)  
Powdered Activated Carbon  
(Sigma Aldrich)



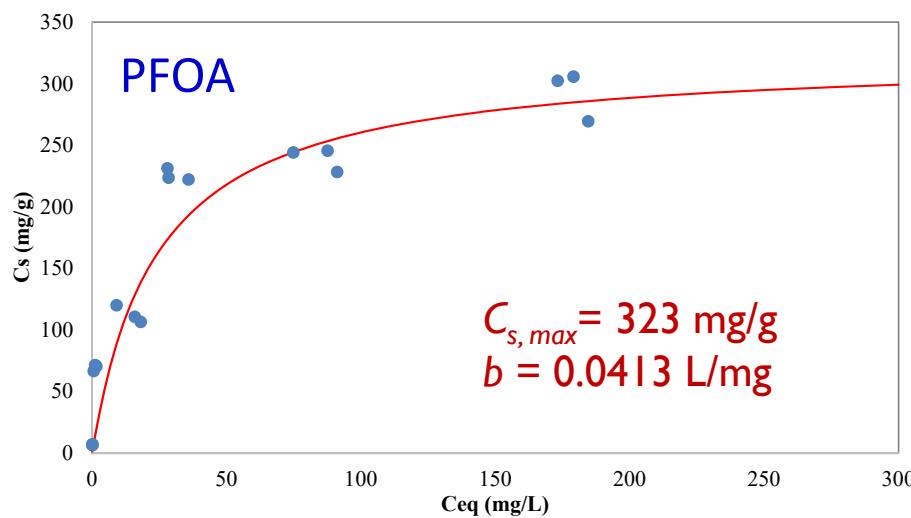
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Provisional Patent Application: Reg. No. 41,942, Docket No. 70011-067P01v (September, 2017)

#ECSUM18

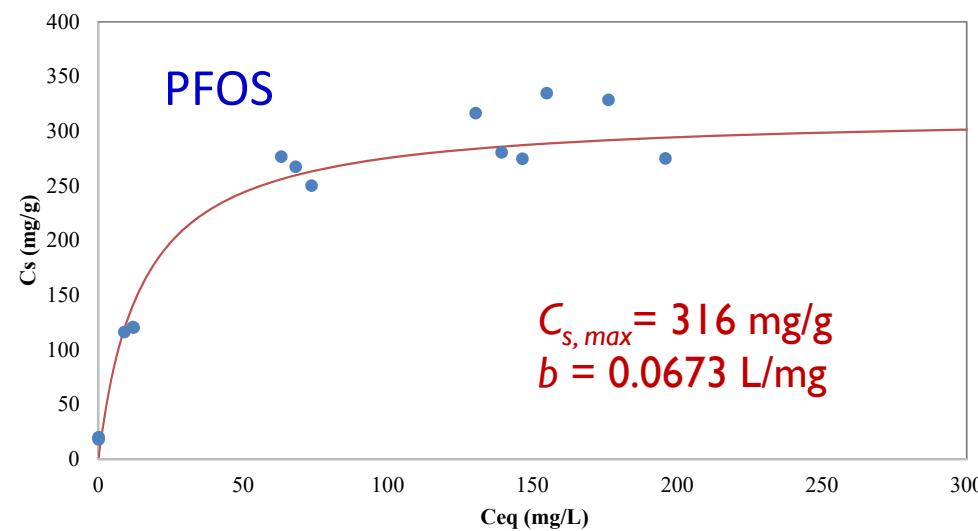
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# PFOA and PFOS Batch Adsorption Studies With Darco® PAC (100-mesh)



Langmuir Isotherm

$$C_s = \frac{C_{s,max} * b * C_{w,eq}}{1 + b * C_{w,eq}}$$



# Injection of PDM+PAC Suspension

$t = 0 \text{ PV}$

Flow  
Direction  
↑



after 3.5 PV PAC+PDM



after 3.5 PV Background



26.8 mg of PAC  
retained in column

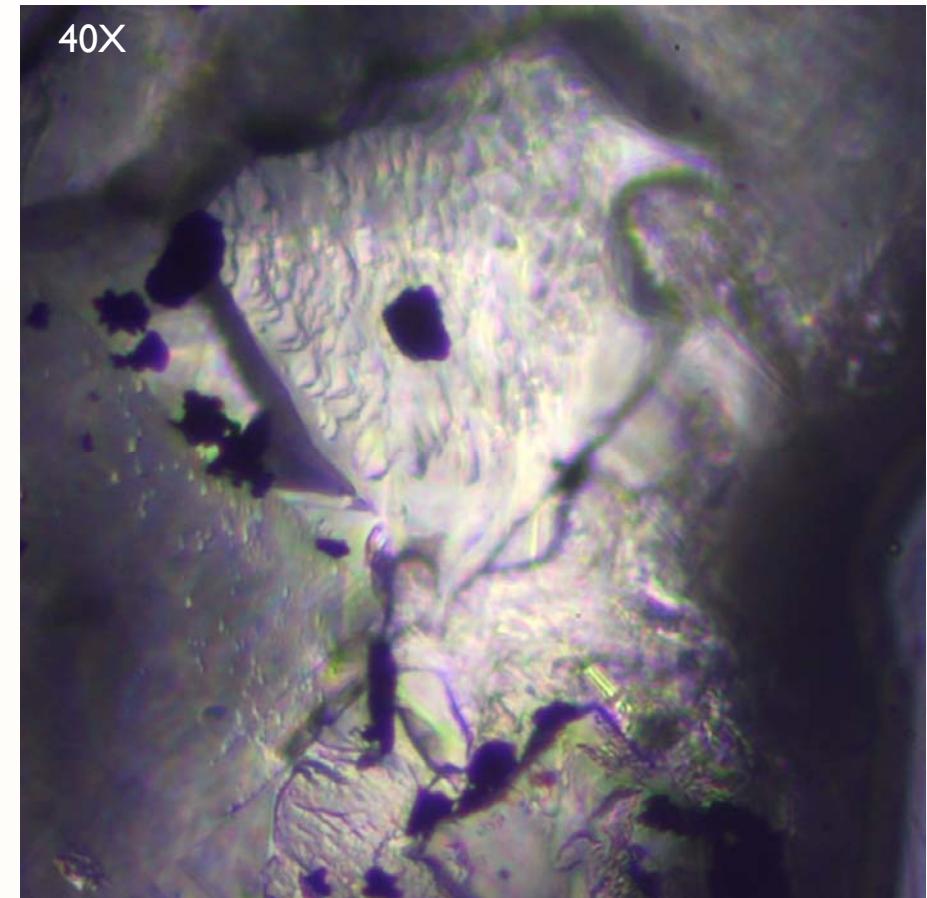
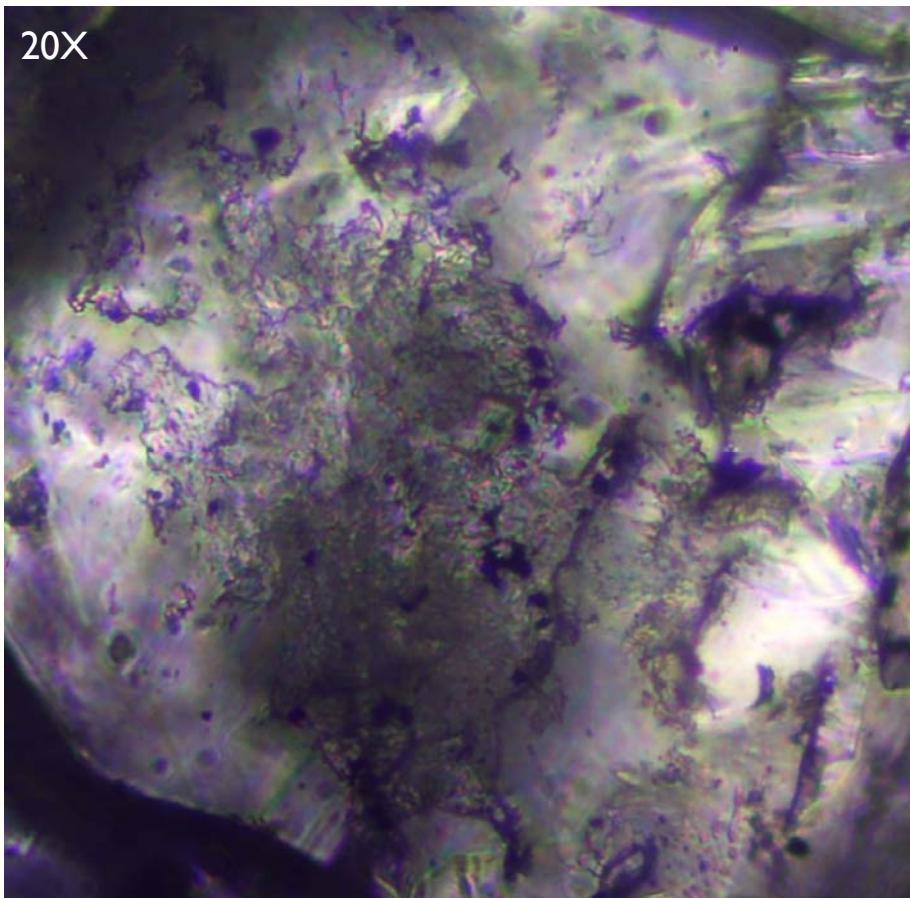
40-50 mesh Ottawa Sand ( $d_{50} = 358 \text{ } \mu\text{m}$ ),  $k_i = 7.37 \times 10^{-11} \text{ m}^2$ ,  $n = 0.37$ , SSA =  $0.0125 \text{ m}^2/\text{g}$ , PV = 22 mL

PDM+PAC Suspension: 1,000 mg/L PAC + 5,000 mg/L PDM, viscosity = 1.18 cP

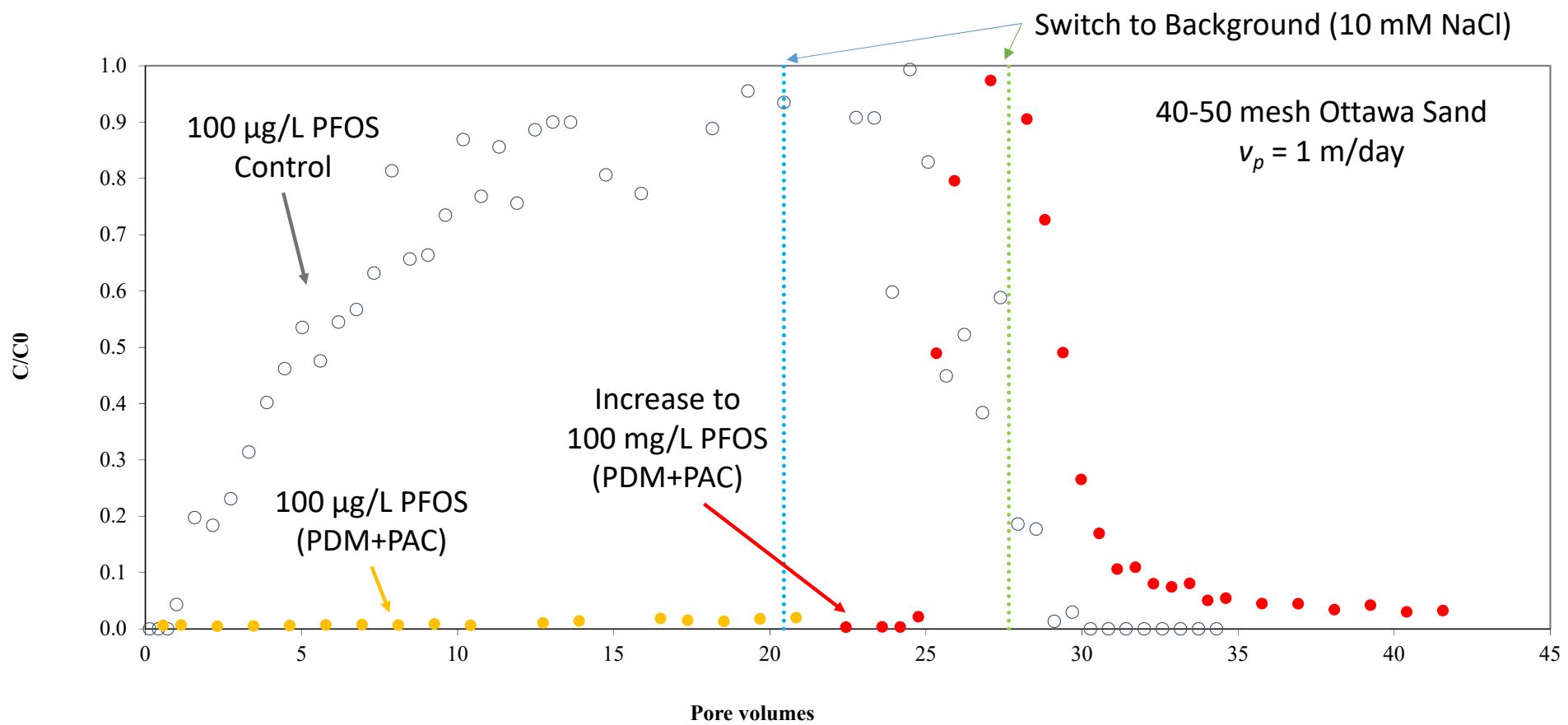
Flow rate: 0.12 mL/min; pore-water velocity ~1.0 m/day

# Images of PDM+PAC Treated Ottawa Sand

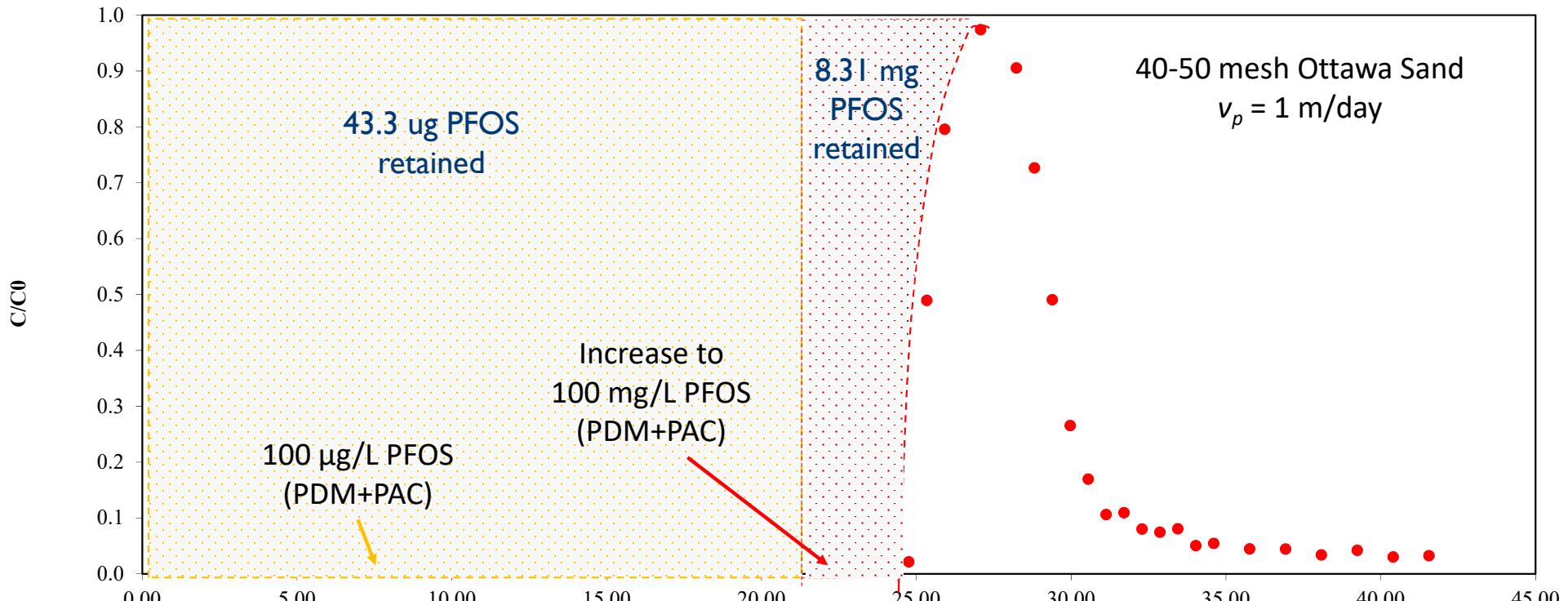
Leica DM IL LED



# PFOS Column: Control; PDM+PAC treated Ottawa Sand

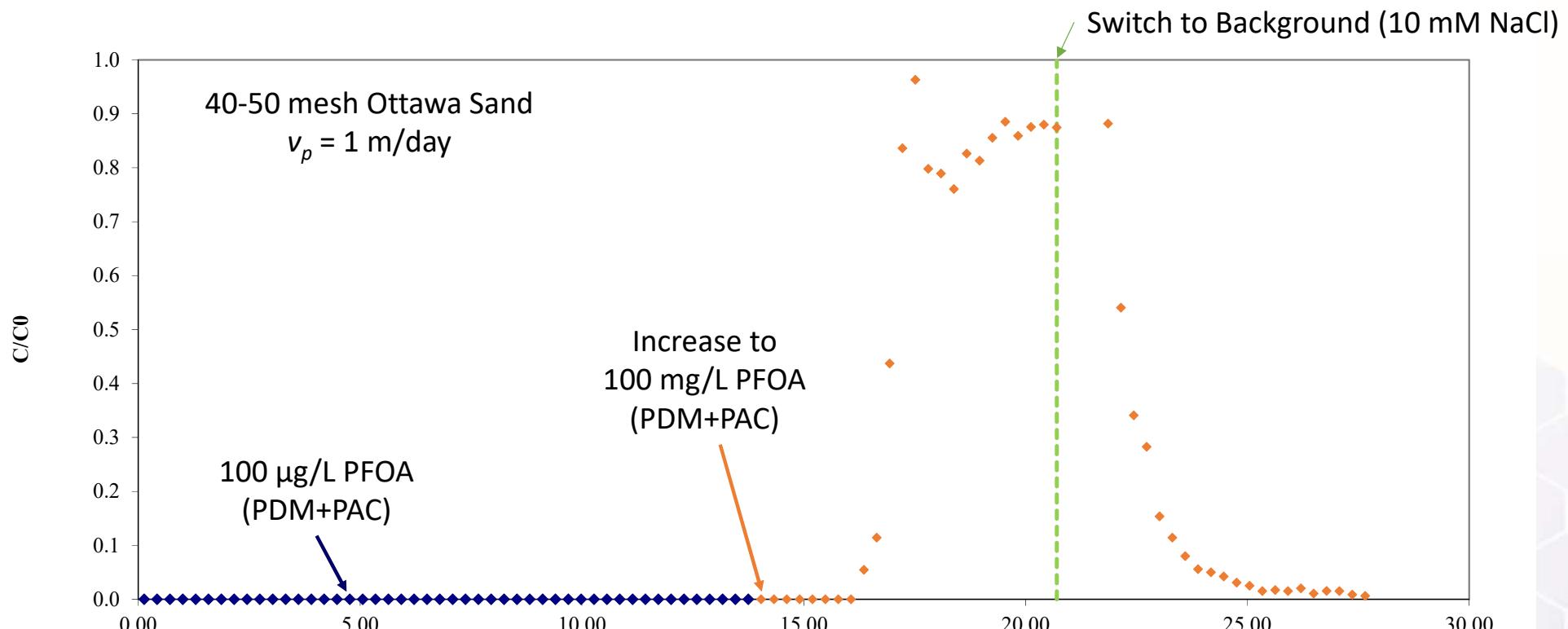


## Retention of PFOS Mass by PDM+PAC Treated Sand



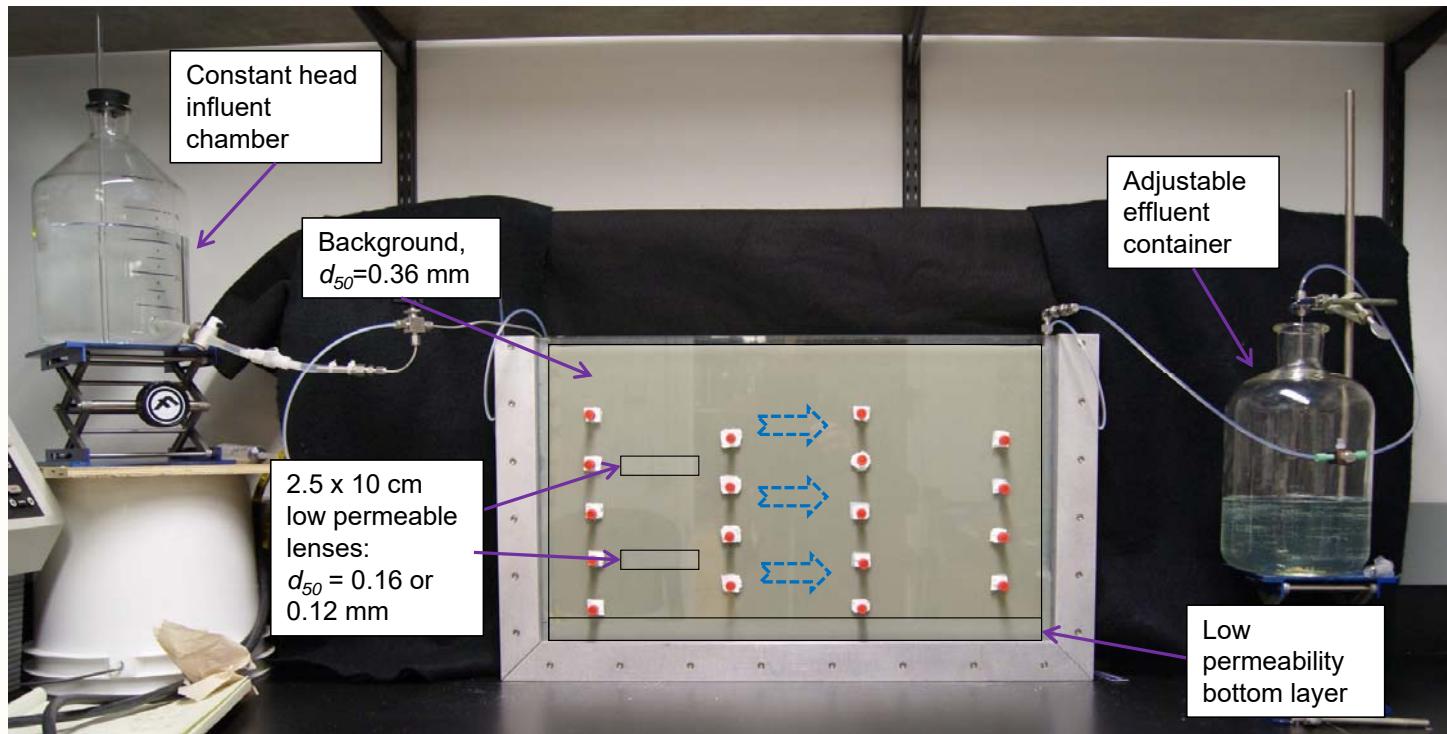
Based on the measured  $C_{s,max} = 316 \text{ mg/g}$  and mass of retained PAC ( $\sim 27 \text{ mg}$ ), the capacity of the column should be  $\sim 8.65 \text{ mg}$  PFOS, consistent with the observed column retention of  $\sim 8.35 \text{ mg}$  PFOS

## PFOA Column: PDM+PAC treated Ottawa Sand



Based on the measured  $C_{s,max} = 323 \text{ mg/g}$  and mass of retained PAC ( $\sim 27 \text{ mg}$ ), the capacity of the column should be  $\sim 8.72 \text{ mg PFOA}$ , consistent with the observed column retention of  $\sim 8.48 \text{ mg PFOA}$

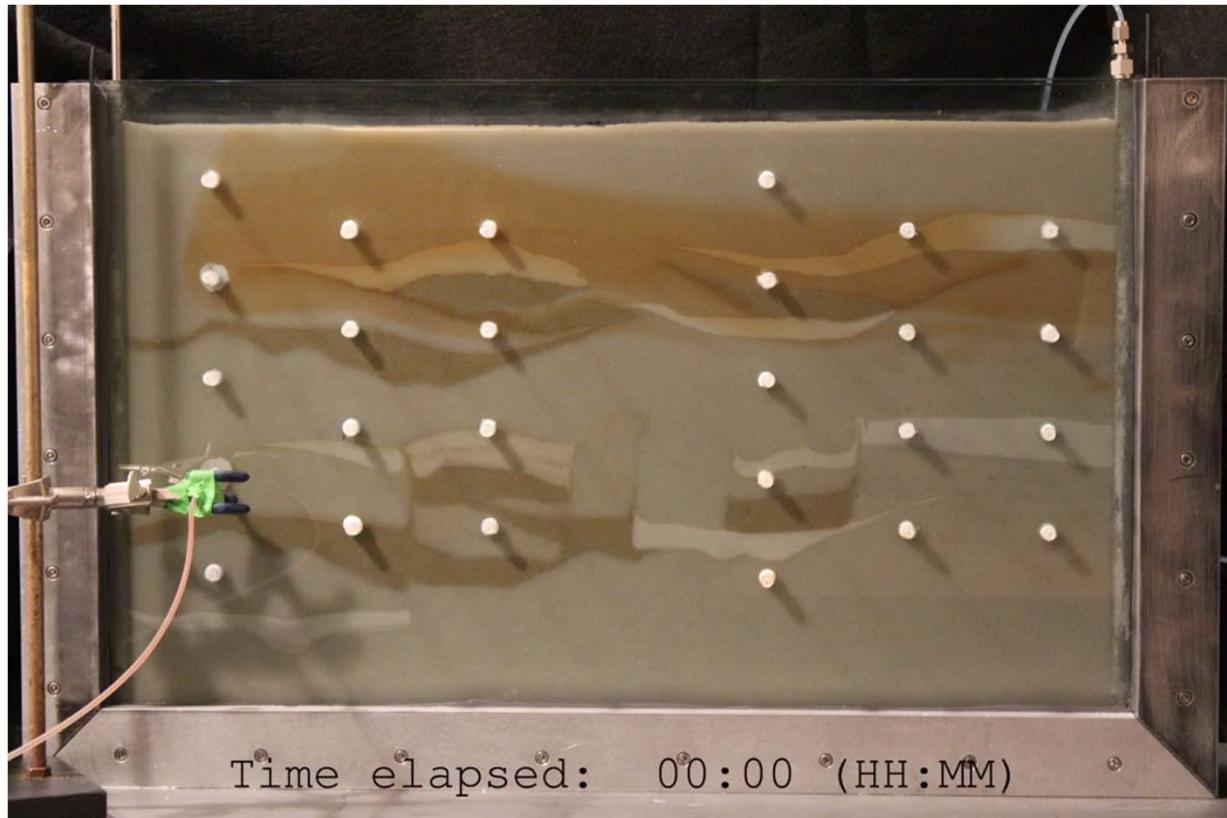
# Heterogeneous 2.5-D Flow Cell



Dimensions: 40.0 cm (ht) x 63.2 cm (length) x 1.4 cm (thickness)



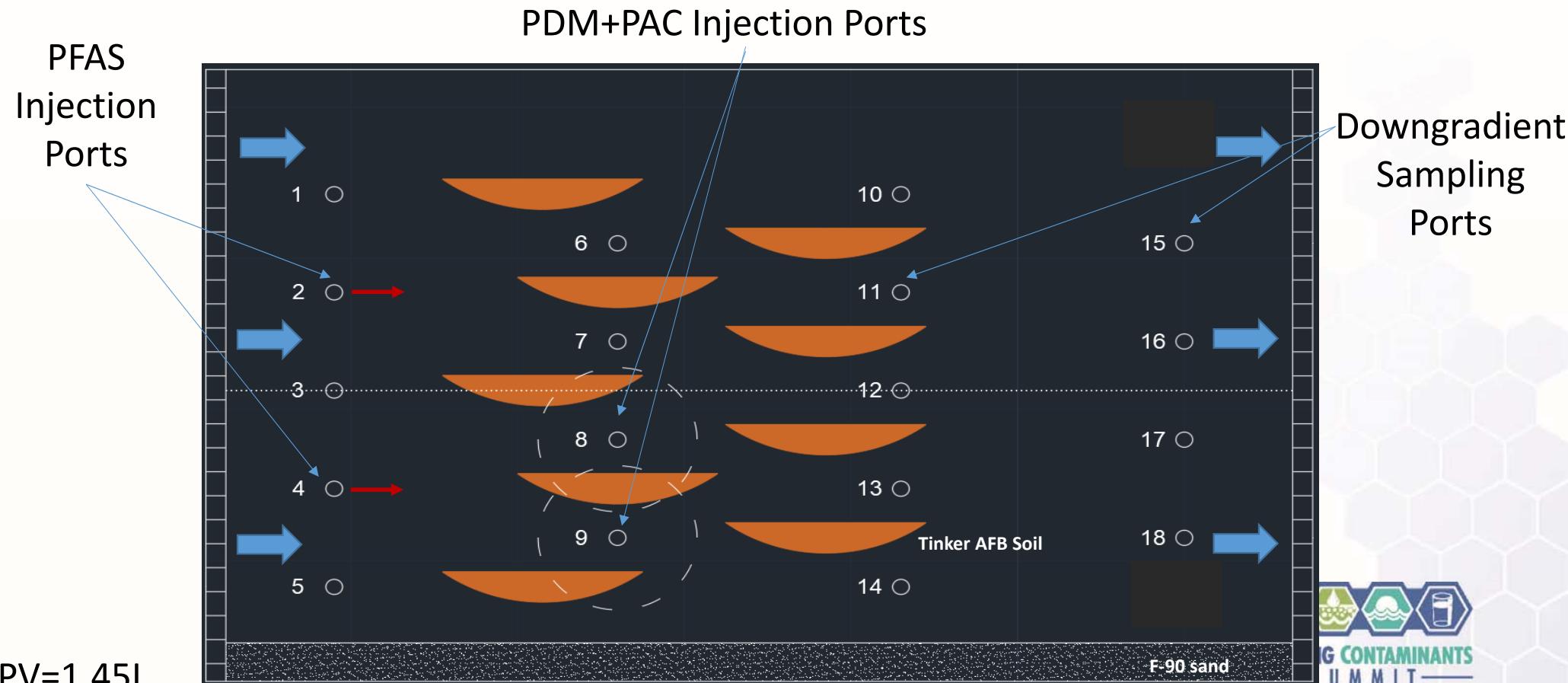
## Injection of Iron Nanoparticles in Heterogeneous Domain



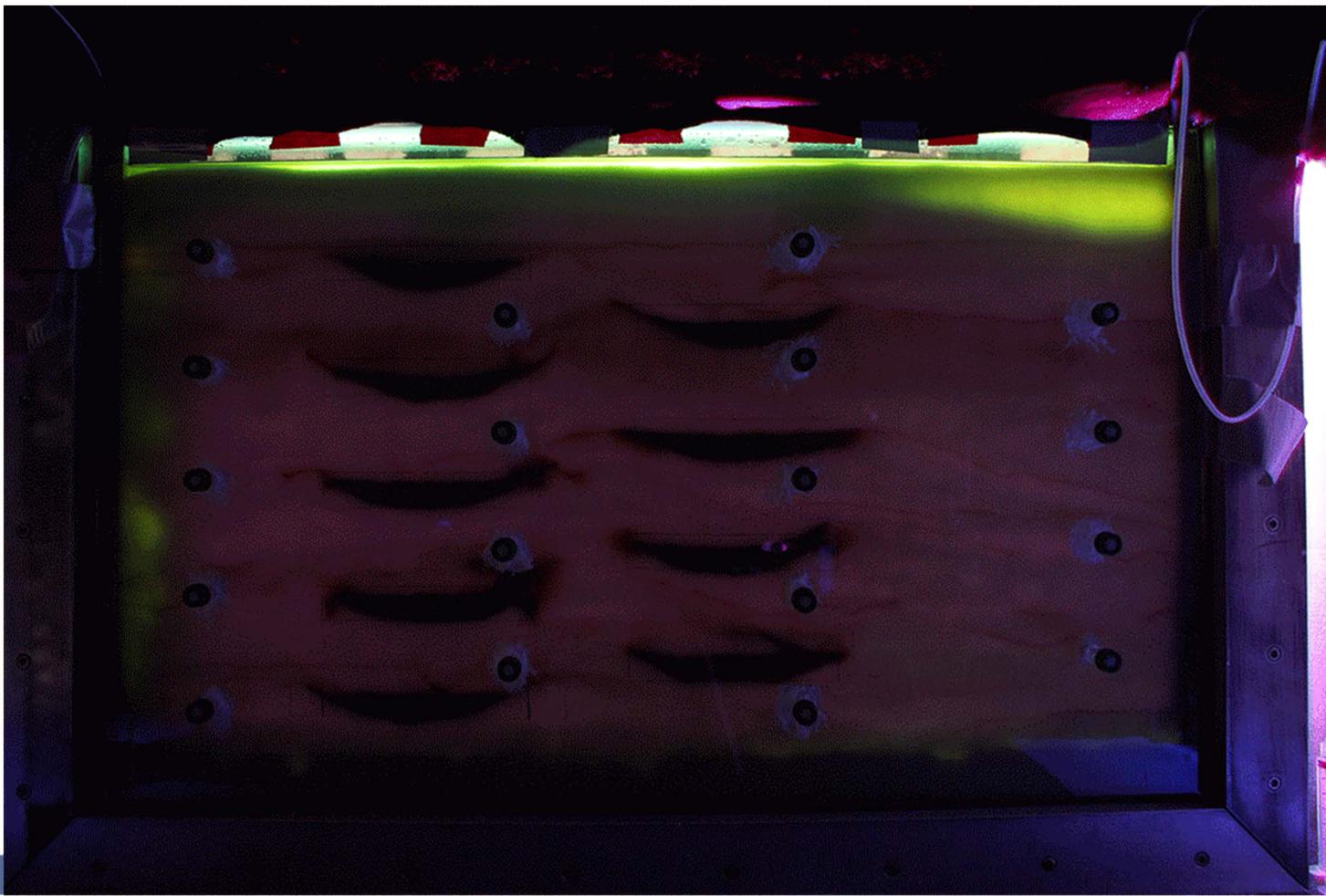
Background velocity = 2 m/d, nMag conc. = 2500 mg/L, gum arabic conc. = 1000 mg/L,  
injected vol. = 100 mL, background = API brine



# Configuration of PFAS 2.5D Flow Cell



## Tracer Test Before PDM+PAC Injection

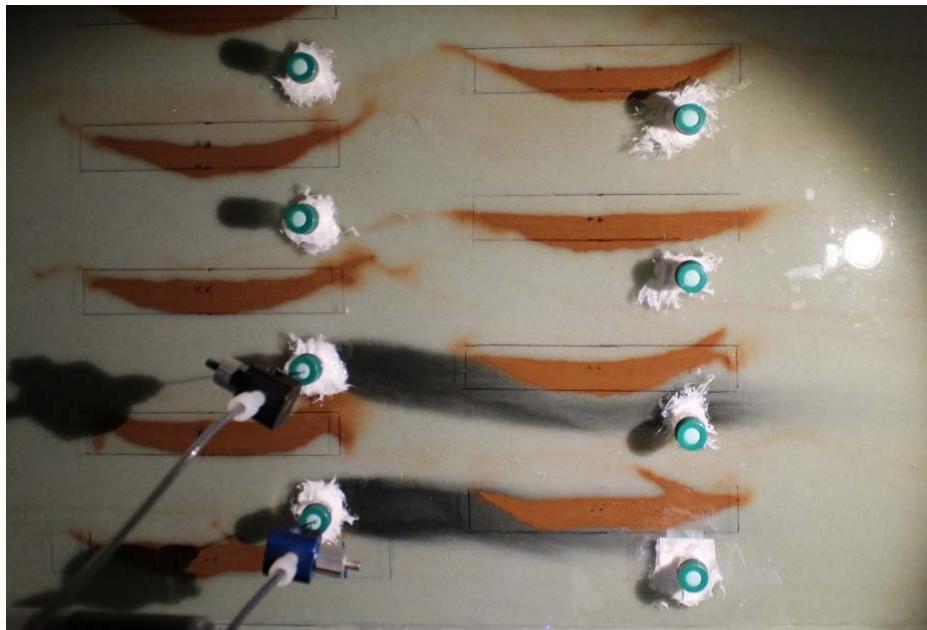


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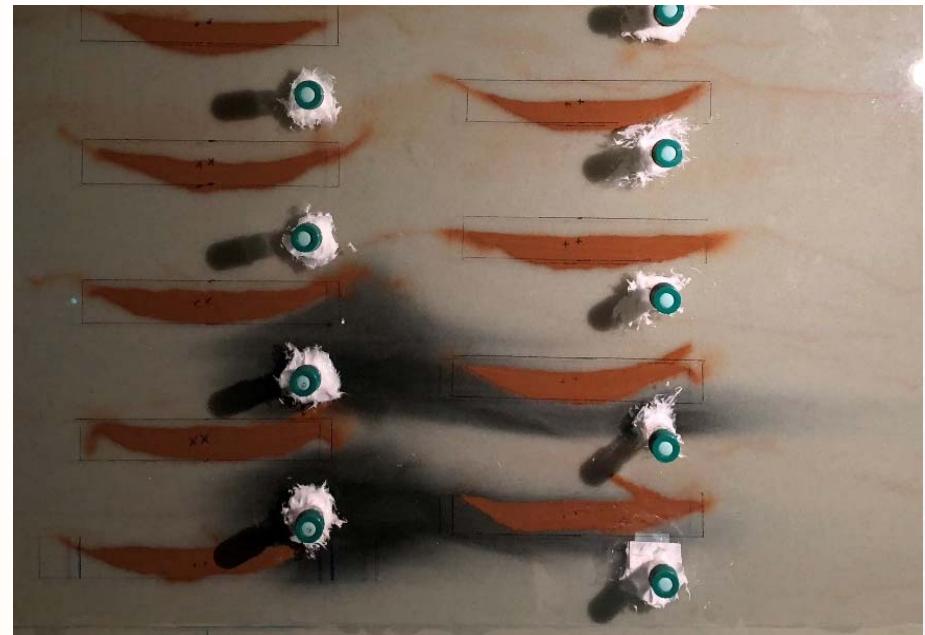
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## Side-port Injection of 1 g/L PAC + 5 g/L PDM

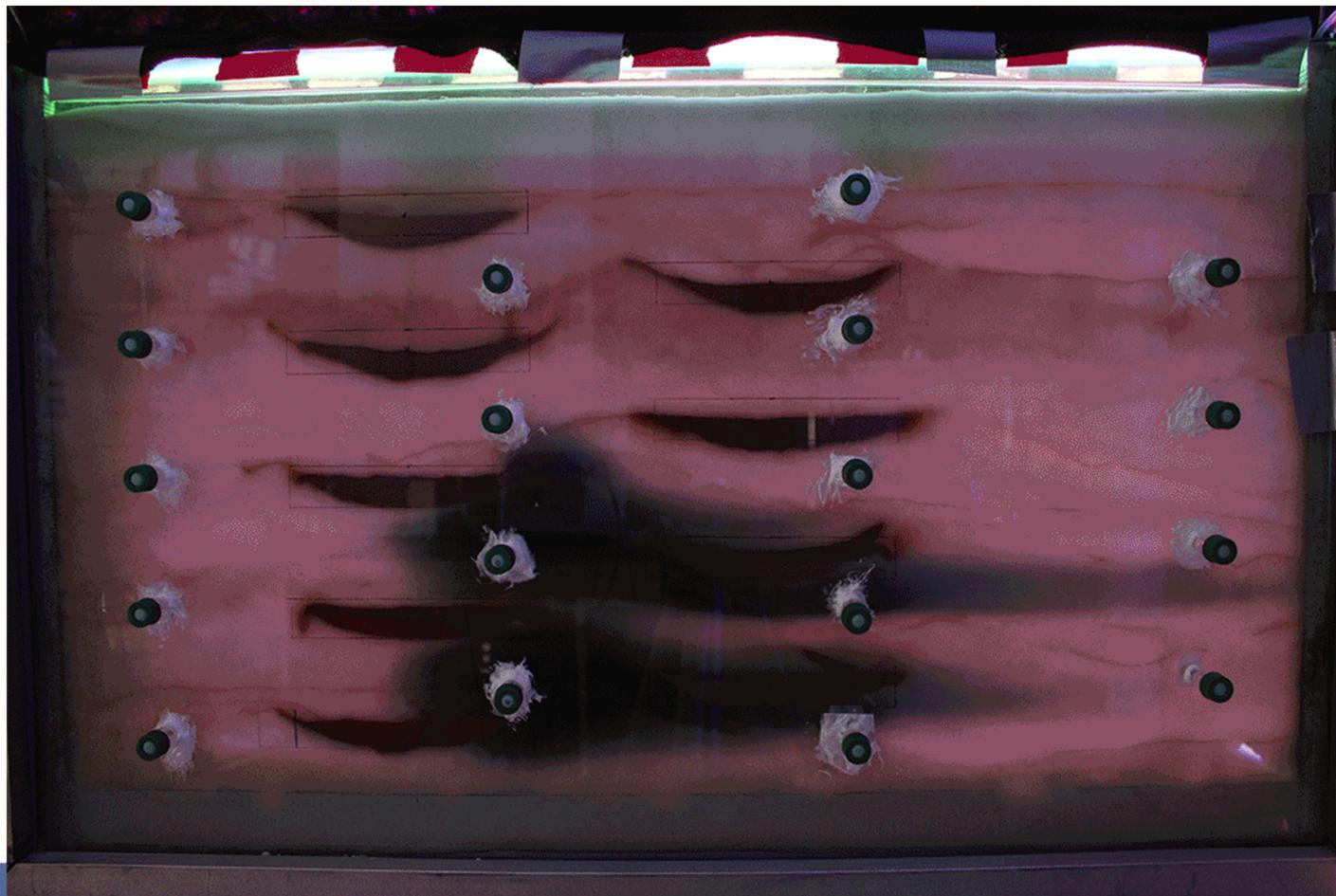
40 mL (0.08 mL/min) with  
background flow (2.4 mL/min)



80 mL (0.08 mL/min) with no  
background flow

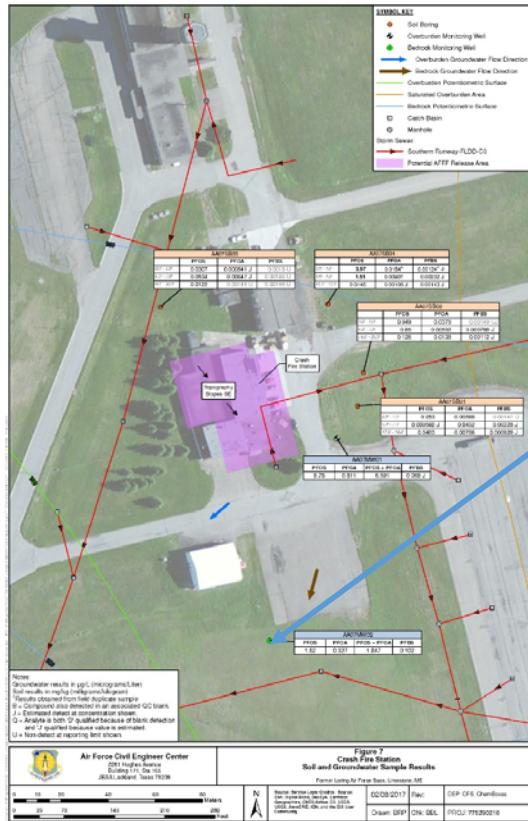


## Tracer Test After PDM+PAC Injection



# Collection of PFAS-Impacted Soil & Groundwater Samples

*Former Loring AFB, Limestone, ME*



AA07MW02		
Compound	Avg (ng/L)	Std. Dev (ng/L)
PFBA	153.01	11.48
PFPeA	377.77	8.56
PFBS	108.36	2.29
PFHxA	632.10	147.02
4:2-FTS	27.54	9.57
PFHpA	147.78	8.08
PFHxS	1341.42	151.11
PFOA	396.86	17.47
6:2-FTS	1145.98	85.72
PFNA	78.42	29.71
PFOS	1604.65	145.16
PFDA	12.20	11.44
8:2-FTS	98.30	26.13



## Conclusions

- Cationic polymers (PDM or PA) increased PFAS sorption by a factor of 3 to 45 based on batch and column experiments.
- The combination of powdered activated carbon (PAC) and PDM formed a stable suspension that can be delivered in situ to form a reactive zone.
- Column and aquifer cell studies demonstrated the sizable capacity of PDM+PAC-treated sand to retain PFOS and PFOA.

## Future Work

- Conduct column and aquifer cell experiments with PFAS mixtures, both laboratory prepared and field groundwater samples (Loring AFB)
- Evaluate the potential release of retained PFAS over time and subject to changing pH and ionic strength.
- Evaluate competitive effects of NOM and other organic contaminants.





Projects ER-2425 and ER-2714



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