Leveraging students' authentic questions about COVID-19 to motivate meaningful engagement with science

mySci is a project of

Washington University in St. Louis

INSTITUTE FOR SCHOOL PARTNERSHIP

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The Institute for School Partnership (ISP) connects Washington University with the surrounding K-12 community to inspire and empower educators and students with the resources they need to succeed.

We identify best practices in teaching and learning and implement these practices in local schools, particularly those with the most vulnerable and underserved students. We translate the most current research in education into learning opportunities for educators at all levels. The Institute for School Partnership Programs



### Best science teaching practices in partnership with Hawthorn Leadership School for Girls



### Implementing Best Practices prior to COVID-19

According to teacher leaders, scientists, and educational researchers, the kinds of experiences that science learners (and teachers) across grades K-12 need and should be able to do:

- Understand, use, and interpret scientific explanations of the natural world
- Generate and evaluate scientific evidence and explanations
- Understand the nature and development of scientific knowledge
- Participate productively in scientific practices and discourse

SOURCES: Taking Science to School: Learning and Teaching Science in Grades K-8 (2007) A Framework for K-12 Science Education Standards (2012) Ambitious Science Teaching (2018)

### Understand, use, and interpret scientific explanations of the natural world

The question we asked: Why does reg	ular sugar make rock candy and Splenda doesn't?	an't make rock candy. (planation B: Sugar can b distance as Splenda I. kno oka chlorine atoms and sple solar dibehave a chara and splenda. I kno very behave differently whe regular atom C: Sugar can b distance as Splenda. The local structure. Sugar colecular structure. Sugar or thave. I know this be ugar and salt and other sul
Explanation	(tape the explanation strip here)	e made into w this becan nda looks m ay. Since ft. F ay. Since ft. nused to m n used to m nused to m splenda and Splenda cause we re cause we re
We think this is the best because		rock candy and Spl se they have differ e molecular comp ade rock candy. Su orco k candy and Spl nore clooks depends are different becat ad about what mak
An Okay Explanation	(tape the explanation strip here)	enda can't becau know they are dif sition of Sugar are sition of Sugar are sup substances plenda conta se spleneda conta es up substances
We think this is just okay because		t can see, like how ferent because Splu d Splenda are differ and splenda doesn't ins an atom that su and we built mode
The Worst Explanation	(tape the explanation strip here)	aame enda he gaa sof
We think this is the worst explanation	because	

# Understand, use, and interpret scientific explanations of the natural world

**Explanation B:** Sugar can be made into rock candy and Splenda can't because sugar is not the same substance as Splenda. I know this because they have different properties that I can see. Ike how sugar looks more grainy and Splenda looks more powdery. I also know they are different because Splenda has chlorine atoms and sugar doesn't. The molecular composition is what causes the properties to look and behave a certain way. Since the molecular composition of Sugar and Splenda are different, they behave differently when used to make rock candy. Sugar forms into it and Splenda doesn't.

- 1. Contains a claim and 2 pieces of evidence.
- 2. Contains strong reasoning that completely answers the question by connecting the unique molecular composition with the unique behavior of Splenda and that of sugar.
- 3. Contains connecting words that increases writing fluency:
  - "I know this because…"
  - Also...
  - Since...
  - Like...

#### The COVID-19 shutdown

- The last day of in-person classes: Tuesday, March 17, 2020.
- Hawthorn moved their Spring Break to **March 23-27** and used this time to prep and plan for distance learning in April.
- **On April 13**, Mary and Heather virtually facilitated a lesson to elicit and discuss students' questions about the COVID-19 pandemic phenomenon.



#### COVID-19 Distance Learning Module



# Generate and evaluate scientific evidence and explanations



#### The scene:

On March 16, 2020, the Imperial College London COVID-response team published <u>a scientific paper</u> that outlined a "worse-case scenario" of 510,000 deaths in the U.K. assuming the country would continue doing nothing to respond to the COVID-19 outbreak. In response to this study, the country (along with the U.S.) made preparations for a public lockdown. As of April 25, 2020, "only" 18,000 deaths have been recorded in the U.K.'s hospitals. That's 28 times fewer than predicted by the initial projections.

- $\oslash$
- 4. When the Imperial College London team revised their initial model-based projections, some news media outlets reported that the team had an error in the initial model. This worried many people that scientists are making mistakes. Some people even argued: we shouldn't trust what these scientists say because their models are wrong. Knowing what you know about models, do you agree or disagree with this statement? Use the CER Checklist on the next page to remind yourself what makes a good argument.

#### Implications and Reflections

- Despite the school shutdowns, we were able to engage students in generating and evaluating scientific evidence and explanations.
- An anchoring phenomenon that is both complex and relevant helped us reframe science from "recreating school at home" to fostering authentic, equitable learning experiences with flexible goals. The design of this unit resulted in higher attendance during synchronous Zoom meetings, more submitted work, and a safe space for students to share how the pandemic had impacted their lives.
- The public, central space for students' questions (i.e. a "driving question board" or "summary table") was instrumental to motivating their subsequent investigations. It is a useful tool to help teachers and students take stock of what they asked, what they did, and what they learned as a result.

#### Questions for Further Investigation

- What if students' questions do not lead to important disciplinary core ideas they're supposed to learn within a unit of instruction?
- How are teachers currently using the coherent kit-based curriculum (mySci) with students?
- How do teachers make decisions about what to adapt and how are they adapting?
- Are teachers and students using the kit materials?