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Competing Polar and Antipolar Structures in the Ruddlesden-Popper Layered Perovskite Li₂SrNb₂O₇

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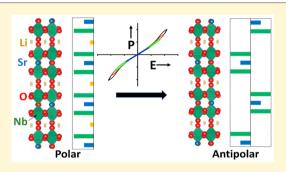
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Supporting Information

ABSTRACT: Over the past few years, several studies have reported the existence of polar phases in n = 2 Ruddlesden–Popper layer perovskites by trilinear coupling of oxygen octahedral rotations (OOR) and polar distortions, a phenomenon termed as hybrid improper ferroelectricity. This phenomenon has opened an avenue to expand the available compositions of ferroelectric and piezoelectric layered oxides. In this study, we report a new polar n = 2 Ruddlesden-Popper layered niobate, Li₂SrNb₂O₇, which undergoes a structural transformation to an antipolar phase when cooled to 90 K. This structural transition results from a change in the phase of rotation of the octahedral layers within the perovskite slabs across the interlayers. First-principles calculations



predicted that the antipolar Pnam phase would compete with the polar A21am phase and that both would be energetically lower than the previously assigned centrosymmetric Amam phase. This phase transition was experimentally observed by a combination of synchrotron X-ray diffraction, powder neutron diffraction, and electrical and nonlinear optical characterization techniques. The competition between symmetry breaking to yield polar layer perovskites and hybrid improper antiferroelectrics provides new insight into the rational design of antiferroelectric materials that can have applications as electrostatic capacitors for energy storage.

1. INTRODUCTION

Noncentrosymmetric materials are interesting to investigate because of their properties of technological relevance, which include piezoelectrics, ferroelectrics, and multiferroics. Ferroelectric oxides are widely used in ultrasonics, electronics, sensors, capacitors, and actuators and are dominated by ABO₃ perovskites, where polarization is induced by off-centering of $3d^0$ cations by second-order Jahn-Teller (SOJT) active effects.¹ This mechanism limits the range of available compositions of acentric perovskites to those that incorporate 3d⁰ cations that can hybridize with O 2p states.² Over the past few years, a series of noncentrosymmetric layered oxides have been predicted by theory and confirmed by experiment, in which inversion symmetry is lifted by a combination of oxygen octahedral rotations (OORs) and A-site cation ordering.³ Benedek and Fennie first proposed a "hybrid improper ferroelectricity" (HIF) mechanism,⁸ where inversion symmetry is lifted by condensation of two distinct symmetry-breaking modes (rotation and tilting of oxygen octahedra) in the n = 2Ruddlesden-Popper (RP) compound Ca₃Mn₂O₇. This

mechanism has since been realized experimentally in several n = 2 RP and Dion-Jacobson (DJ) phase layer perovskites^{7,9–17} as well as in perovskite superlattices^{18,19} and molecular perovskites.²⁰ In related work, Rondinelli and co-workers employed group theoretical methods to elucidate the crystal chemical principles that govern the design of acentric n = 1 RP $AA'BO_4$ layer perovskites.²¹ Over the past few years, a number of layered oxides with polar space group symmetry have been reported experimentally, and noncentrosymmetricity has been attained by the trilinear coupling of A-site cation displacements and OORs.^{7,10,22}

More recently, the HIF mechanism has been extended to include antiferroelectric displacements of A-site cations driven by a reversal in the direction of OORs. Antiferroelectrics are materials that contain electric dipoles oriented antiparallel within individual domains. They are known to have low

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dielectric loss and are gaining attention as energy storage materials, but they cannot be readily distinguished from other centrosymmetric materials by symmetry.²³ Another limitation is that most known antiferroelectrics are lead-based compounds that are toxic. One approach to overcome these limitations is to search for centrosymmetric materials that have polar metastable phases. Lu et al. predicted a polar-to-nonpolar phase transition in a series of n = 2 RP A₃B₂O₇ thin films resulting from compressive and/or axial strain, which could be used to tune weak ferromagnetism in antiferromagnetic compounds by varying the applied electric field at the boundary of the phase transition.^{24,25} Similarly, Nowadnick et al. identified the antipolar Pnam phase as one of the ferroelectric switching pathways in the $n = 2 \text{ RP } \text{Ca}_3 \text{Ti}_2 \text{O}_7$ using group theory, where the polar and antipolar phases differ merely in the octahedral rotation sense $(a^0a^0b^+)$ across the interlayers.²⁶ Recently, Yoshida et al. observed a structural transformation from a polar $A2_1am$ phase to a nonpolar Pnab phase by an in-phase to out-of-phase change in the OOR sense in n = 2 RP Sr₃Zr₂O₇.¹¹ This kind of polar-to-antipolar phase transition could be very interesting for realizing new antiferroelectric materials for energy storage applications.

In this study, we report another polar-to-antipolar phase transition in a polycrystalline n = 2 RP phase layered niobate Li₂SrNb₂O₇, which contains intergrowths of two perovskite SrNb₂O₆ layers with Li–O interlayers when cooled from room temperature to 90 K. The structure of Li2SrNb2O7 was previously assigned to the centrosymmetric Amam space group,²⁷ but this study reveals that the polar A21am and antipolar Pnam structures are energetically more stable by both theory and experiment. The two structures result from the coupling of two octahedral rotations and tilts (denoted by irreducible representations (irreps) X_3^- and X_2^+) along different crystallographic axes and differ merely in the phase of their rotation across the interlayers, i.e., order parameter directions of the irreps. In contrast to studies on A₃B₂O₇ systems that mentioned that the nonpolar *Pbcn* phase $(a^{-}a^{-}c^{-})$ competes with the polar $A2_1am$ phase, this study reveals that there is a competition between the polar A21am and the antipolar Pnam phases, both of which have OORs denoted by $a^{-}a^{-}c^{+}$. In the former structure, the combination of the $X_3^-(a, -a)X_3^-(a, -a)$ and $X_2^+(a, -a)X_2^+(a, -a)$ distortion modes lifts inversion symmetry and induces a polar $\Gamma_5^-(a, a)$ distortion mode, while in the latter structure, the combination of the $X_3^-(a, -a)X_3^-(a, -a)X_3^$ -a) and the $X_2^+(a, a)X_2^+(a, -a)$ distortion modes keeps inversion symmetry and induces an antipolar $M_5^-(a, 0)$ distortion mode. A phase diagram of Li2SrNb2O7 was determined by combining optical second harmonic generation, temperature-dependent synchrotron X-ray and neutron diffraction, and electrical characterization.

2. EXPERIMENTAL SECTION

2.1. First Principles Calculations. First-principles calculations based on density functional theory (DFT) were carried out using the projector augmented-wave (PAW) method^{29,30} and the GGA-PBEsol functional^{31–33} as implemented in the Vienna Ab initio Simulation Package (VASP) code.^{34–37} [Certain commercial equipment, instruments, or materials are identified in this document. Such identification does not imply recommendation or endorsement by the National Institute of Standards and Technology nor does it imply that the products identified are necessarily the best available for the purpose.] PAW data sets with radial cutoffs of 1.4 Å, 2.1 Å, 1.5 Å, and 0.8 Å were employed for Li, Sr, Nb, and O, respectively. Li 1s, Sr 4s, 4p, and 5s, Nb 4p, 4d, and 5s, and O 2s and 2p orbitals are treated as valence

electrons. A plane-wave cutoff energy of 550 and 700 eV was used when structural optimization and density functional perturbation theory (DFPT) calculations were performed, respectively. Γ -centered k-point mesh sampling of 4 × 4 × 1 was used for the $\sqrt{2}\times\sqrt{2}\times1$ supercells of the conventional *I4/mmm* unit cell with 24 atoms, except for the DFPT calculations, where an 8 × 8 × 2 mesh was used for the accuracy. Lattice constants and internal atomic coordinates were optimized until residual forces and stresses become less than 0.01 eV/Å and 0.08 GPa, respectively. The Born effective charge tensors were calculated by using DFPT calculations implemented in the VASP code.

Phonon frequencies were derived from the calculated force constants using the PHONOPY code.³⁸ The phonon dispersion curves were displayed for a standard primitive cell^{39,40} along a q-space path based on crystallography.⁴¹ Ground-state structures for Li₂SrNb₂O₇ were explored via a method described by Togo and Tanaka⁴² as follows: zone-center phonon frequencies were calculated for the $\sqrt{2} \times \sqrt{2} \times 1$ supercell of the conventional cell of the aristotype I4/mmm phase. After perturbing the structure according to the eigenvectors of the obtained imaginary-frequency unstable phonon modes, structure optimizations were performed under symmetry constraints to obtain new stable structures with lower symmetry. Subsequently, we calculated phonon frequencies for the fully relaxed structures to see if unstable modes were still found. If not found, the structure was proven to be dynamically stable. If found, structural perturbations and subsequent optimization were done. This process was repeated until no unstable phonon modes were observed. The cell size was fixed during the structure exploration. We note that zone-boundary phonon modes at the X and M points, which are (1/2)1/2 0) and (0 0 1) in the conventional basis ((0, 0, 1/2) and (1/2, 1/ 2, -1/2) in the primitive basis),⁴³ respectively, are folded into the Γ point for the $\sqrt{2} \times \sqrt{2} \times 1$ supercell. Therefore, we explored stable structures within a subspace spanned with linear combinations of atomic displacement modes at the Γ , X, and M points. This method has been applied to ground-state structure explorations for other layer perovskite systems, resulting in a good agreement with the results of experimental structural identifications.^{13,14} Symmetry mode analysis was performed to decompose distortions into symmetry-allowed modes with relevant irreps using the AMPLIMODES tool.44,45

2.2. Synthesis. Polycrystalline Li₂SrNb₂O₇ was synthesized by grinding the precursors Li₂CO₃ (>99%, Aldrich), SrCO₃ (>99.9%, Aldrich), and Nb₂O₅ (>99.5%, Alfa Aesar) and heating them at 600 °C for 12 h followed by regrinding, pelletizing, and heating the pellet at 1050 °C for 6 h. The second heating step was repeated until the phase-pure compound was obtained.

2.3. Materials Characterization. Phase purity was confirmed by laboratory X-ray diffraction using a Malvern Panalytical theta-theta X-ray diffractometer (monochromatized Cu K α radiation). Temperature-dependent synchrotron X-ray diffraction patterns (SXRD) were obtained at the 11-BM-B beamline of Argonne National Laboratory $(\lambda = 0.412632 \text{ Å})$ using 12 Si (111) crystal analyzers and LaCl₃ scintillation detectors. Neutron powder diffraction (NPD) was performed at the BT-1 powder diffractometer at NIST using Ge(311) and Ge(733) monochromators with wavelengths of 2.079 and 1.197 Å, respectively. Measurements were performed at 90 and 300 K after loading the sample in a vanadium can. Rietveld refinement of synchrotron X-ray and neutron diffraction patterns were performed by using GSAS-EXPGUI programs.^{46,47} Temperature-dependent optical SHG measurements were performed in reflection geometry using a Ti-sapphire laser of incident wavelength 800 \pm 20 nm and 80 fs pulses at a frequency of 2 kHz. Polarization-field hysteresis and current versus field measurements were performed by first depositing 200 nm Pt electrodes on a dense pellet for electrical contact and then applying a DC field of 1 kV-10 kV using a Sawyer-Tower circuit at a measurement frequency of 10 Hz. Low temperature dielectric measurements were obtained using a Precision LCR Meter (HP4284A, Agilent Technologies, Santa Clara, CA) at a frequency range of 10³-10⁶ Hz from 300 K down to 100 at 10 K intervals.

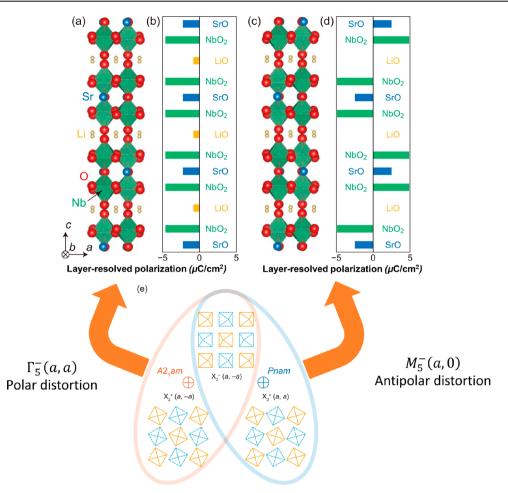


Figure 1. (a) Calculated crystal structure and (b) layer resolved polarization of n = 2 RP Li₂SrNb₂O₇ in the polar $A2_{1}am$ space group. (c) Calculated crystal structure and (d) layer resolved polarization of n = 2 RP Li₂SrNb₂O₇ in the antipolar *Pnam* space group. (e) Schematic of octahedral rotation patterns from which the $A2_{1}am$ and *Pnam* structures are constructed with fixed tilt. The blue and orange octahedra represent alternating layers and are corner-shared but their connectivity is omitted to clearly show the rotation patterns.²⁸

3. RESULTS AND DISCUSSION

3.1. First Principles Calculations. To check dynamical stability, phonon dispersion curves were calculated for the

Table 1. Calculated Energies of Li₂SrNb₂O₇ with Different Oxygen Octahedral Rotations (in meV/formula unit)

	space group	OOR pattern ^a	total energy ^b (meV/f.u.)	comments	
	Amam	$(\phi\phi 0)$	-81	previously reported ²⁷	
	$A2_1 am$	$(\phi\phi\psi_z)$	-89	polar structure	
	Pnam	$(\phi\phi\psi_z)(\phi\phi\overline{\Psi_z})$	-90	antipolar structure	
	Стса	$(00\psi_z)$	0		
а	Octahedra	l rotation natterns	are in Aleksand	rov notation ^b Total	

"Octahedral rotation patterns are in Aleksandrov notation. "Tota energies are with respect to the parent *I*4/*mmm* structure.

aristotype *I4/mmm* phase (Figure S1). Phonon dispersion curves showed structural instabilities at the Γ , X, and M points that were frequently observed to generate lower symmetry structures in other RP systems.^{13,14} To thoroughly search for the most stable structural phase by unstable phonon modes based on the calculations, a "family-tree" diagram was constructed that lists the total energies of all the structures with respect to the parent *I4/mmm* structure (Figure S2).

First-principles calculations by DFT revealed that, among the several metastable phases, the polar A21am phase was energetically similar to the antipolar Pnam phase and both had the lowest energy with respect to the parent I4/mmm phase (Figure S2). The two most stable structures of Li₂SrNb₂O₇ are the antipolar *Pnam* structure whose net polarizations in alternating double perovskite layers are canceled and the polar A21am structure which has net polarization (Figure 1). One of the common maximum supergroups of the two phases is the Amam space group. The corresponding Amam structure has an $a^-a^-c^0$ OOR pattern. Symmetry mode analysis for the two structures with respect to the Amam phase revealed that the polar A21am and *Pnam* phases are derived by $\Gamma_2^-\Gamma_2^-$ and $Y_2^-Y_2^-$ distortion modes from the Amam phase, respectively. Symmetry mode analysis with respect to the parent I4/mmm structure provides us with more insight into the symmetry aspects of the structural features. Both the structures have the same OORs represented by the irreps, X_2^- and X_3^+ (Table S1). The difference is in the direction of the order parameter of the irreps. Depending on the relative direction of the X_2^- and X_3^+ modes, induced modes can be polar (Γ_5^-) or antipolar (M_5^-) . In the A2₁am structure, the combination of the $X_2^-(a, -a)$ and $X_3^+(a, -a)$ distortion modes breaks inversion symmetry and induces a polar $\Gamma_5^-(a, a)$ distortion mode. As in hybrid improper ferroelectrics, the

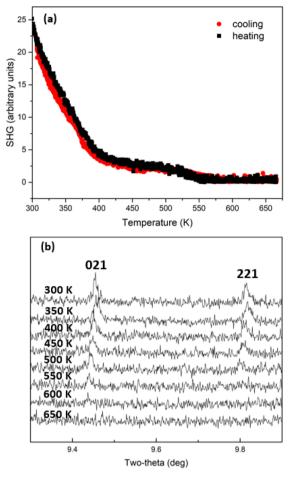


Figure 2. (a) Temperature-dependent SHG measurement of polycrystalline $Li_2SrNb_2O_7$. (b) Tracking the disappearance of the 021 and 221 superlattice reflections with increasing temperature using synchrotron X-ray diffraction.

freezing of two types of OORs drives the polar distortion. On the other hand, in the Pnam structure, the combination of the $X_2^-(a, -a)$ and $X_3^+(a, a)$ distortion modes induces an antipolar $M_5^-(a, 0)$ distortion mode, retaining inversion symmetry. The freezing of two types of OORs leads to the antipolar distortion. Thus, the Pnam phase could be a hybrid improper antiferroelectric.^{13,24} When the directions of tilts are fixed and rotations are varied (Figure 1 e), the two structures have the same rotation direction for the two layers within a perovskite block, but the direction is in-phase across neighboring double perovskite layers. The condensation of the $X_3^-(a, -a)X_3^-(a, -a)X_3^-(a,$ (-a) distortion mode leads to the common supergroup Amam phase. The $\Gamma_2^{-}\Gamma_2^{-}$ and $Y_2^{-}Y_2^{-}$ distortion modes with respect to the Amam phase mainly consist of the combination of the $X_2^+(a, -a)X_2^+(a, -a)$ and $\Gamma_5^-(a, a)\Gamma_5^-(a, a)$ distortions and that of the $X_2^+(a, a)X_2^+(a, a)$ and $M_5^-(a, 0)M_5^-(a, 0)$ distortions, respectively (Figure S3). The difference in the rotation patterns for A2 1am and Pnam structures can be represented by the Glazer notation $a^-a^-c^+/a^-a^-c^+$ and $a^-a^-c^+/a^-a^--(c^+)$ as described before⁴⁸ and can be more easily distinguished using Aleksandrov notation (Table 1).

3.2. Structural Characterization. Optical second harmonic generation (SHG) showed a finite signal at 300 K, consistent with the polar $A2_1am$ structure. The gradual disappearance of the SHG signal at 550 K indicates a transition to a centrosymmetric structure, which was found to be

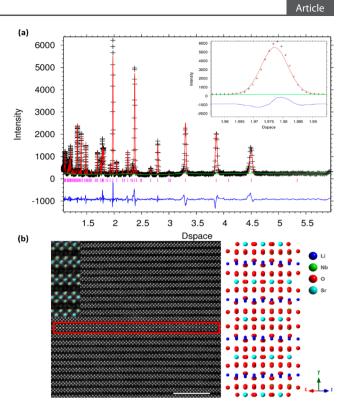


Figure 3. (a) Structural refinement of Li₂SrNb₂O₇ neutron diffraction data using the $A2_1am$ model at 300 K ($\chi^2 = 4.51$). The inset shows the enlarged view of the (511) reflection, the profile of which is largely asymmetric. (b) High-angle annular dark-field scanning transmission electron microscopy of Li₂SrNb₂O₇ along the [101] zone axis showing a stacking fault highlighted by a red box. The inset shows the mapping of Sr and Nb atoms within the layers that match with the simulated structure on the right. Scale bar is 5 nm.

reversible upon cooling (Figure 2a). At low temperatures, SHG shows a thermal hysteresis behavior where the signal peaks at 150 K but does not recover until 200 K during the heating cycle, thus indicating a first-order phase transition (Figure S4). Room temperature synchrotron X-ray diffraction patterns revealed the presence of two weak (021) and (221) superlattice reflections (i.e., $\frac{1}{2} \frac{3}{2} 0$ and $\frac{1}{2} \frac{3}{2} 2$ with respect to the I4/mmm structure) that correspond to $\sqrt{2a_p} \times \sqrt{2a_p} \times c_p$ enlargement of the unit cell with respect to the aristotype tetragonal I4/mmm structure (Figure 2b). The room temperature scan matched well with both the A21am and the Amam structures, but they could not be readily distinguished using Xrays (Figure S5). On the other hand, powder neutron diffraction at 300 K indicated that the room temperature phase could be fitted to the polar $A2_1am$ structure (Figure 3a), which gave a slightly better fit than the Amam structure (Figure S6). Thus, the structure was found to be polar at room temperature, in good agreement with the SHG result. The refinement of both synchrotron X-ray and neutron diffraction data was complicated by the presence of stacking faults, which cause largely asymmetric peak profiles (the inset of Figure 3a). The stacking fault could also be imaged by high angle annular dark field scanning transmission electron microscopy (HAADF-STEM) imaging along [101] axis which showed a triple layered perovskite block as highlighted (Figure 3b). Both HRTEM and HAADF-STEM images indicated multiple kinds of stacking faults but did not accurately reveal all the different types of stacking faults present (Figure S7). Tables 2 and 3 list

atom	Wyckoff position	x	у	z	$U_{ m iso}$
Li1	8b	-0.5100(0)	-0.0038(3)	0.7598(6)	0.0000(5)
Nb1	8b	-0.7239(0)	0.2485(2)	0.3845(6)	0.0006(0)
O1	8b	-0.4855(7)	0.0043(1)	0.5959(6)	0.0037(3)
02	8b	0.0155(6)	0.4934(7)	0.3835(7)	0.0054(0)
O3	8b	-0.7354(8)	0.2202(2)	0.2823(7)	0.0045(0)
O4	4a	-0.7412(5)	0.7874(1)	0.00000	0.0055(5)
Sr1	4a	-0.7369(9)	0.2452(6)	0.00000	0.0046(9)

Table 2. Refined Atomic Positions and Thermal Parameters for $Li_2SrNb_2O_7$ against the $A2_1am$ Model Using SXRD Data at 300 K^a

"Values in parentheses indicate one standard deviation. Lattice parameters: a = 5.5918(5) Å, b = 5.5987(0) Å, and c = 18.0102(6) Å and $R_{wp} = 31.88\%$ and $R_p = 7.75\%$.

Table 3. Refined Atomic Positions and Thermal Parameters for $Li_2SrNb_2O_7$ against the $A2_1am$ Model Using NPD Data at 300 K^a

atom	Wyckoff position	x	у	z	$U_{ m iso}$
Li1	8b	-0.4659(9)	0.0259(9)	0.7615(9)	-0.0171(9)
Nb1	8b	-0.7385(2)	0.2462(2)	0.3844(9)	0.0046(9)
O1	8b	-0.4950(2)	0.0032(3)	0.5981(5)	0.0002(8)
O2	8b	0.0224(3)	0.5030(8)	0.3838(9)	0.0064(5)
O3	8b	-0.7673(8)	0.2235(1)	0.2810(9)	0.0052(4)
O4	4a	-0.7725(8)	0.7789(1)	0.00000	0.0058(0)
Sr1	4a	-0.7606(2)	0.2562(2)	0.00000	0.0050(1)

Values in parentnesses indicate one standard deviation. Lattice parameters: a = 5.5932(0) A, b = 5.5910(8) A, and c = 18.0001(9) A and R_{wp} 10.08% and $R_p = 7.84\%$.

the refined atomic coordinates for $Li_2SrNb_2O_7$ using SXRD and NPD data, respectively. The refined structure for the $A2_1am$ phase showed a small structural distortion as indicated by small OOR (Figure S8).

At high temperatures, SXRD showed the disappearance of the (021) and (221) superlattice reflections at 650 K and revealed a phase transition to the tetragonal *I*4/*mmm* structure. The disappearance of the (221) reflection at 550 K likely corresponds to a centrosymmetric structure as it correlated well with the disappearance of SHG signal, but the space group could not be determined unambiguously. The disappearance of the (221) reflection did not match the Amam model (that is expected from first principles calculations) as it has the same reflection conditions as A21am structure. With decreasing temperature, several new reflections appeared at 200 K that matched well with the antipolar Pnam structure. These include reflections such as (141), (270), (290), (341), (490), (690), and many more that are forbidden in the $A2_1am$ space group, as they do not obey the reflection condition k + l = 2n for Acentered lattices. Figure 4a,b tracks the thermal evolution of the (141) and (341) reflections with temperature and reveals that there is a marked increase in the intensity of these reflections at around 200 K. However, in these temperature ranges, there seems to be a coexistence of both polar and antipolar phases, thus indicating a diffuse phase transition between the A21am and the Pnam phases. Powder neutron diffraction collected at 90 K could be fitted with the Pnam model (Figure S9 and Table S2).

To further elucidate the electrical properties of the emergent phase at low temperatures, polarization-field hysteresis loops, P(E), were measured under liquid nitrogen. The P(E)measurements suggested antiferroelectric behavior at 77 K as shown by the double hysteresis loops with polarization values of $\approx 1.0 \ \mu\text{C/cm}^2$, although the loops could not be saturated (Figure 4c). The measured polarization was much lower than the calculated value by DFT ($25 \ \mu C/cm^2$). The low polarization value was attributed to the polycrystalline nature of the sample which had random orientations and was consistent with previous polarization values obtained on polycrystalline samples.¹³ Additionally, the current (*I*) vs voltage (*V*) curve showed a peak in the current at high electric fields resulting from polarization that is suggestive of antiferroelectric behavior (Figure S10). Room temperature *P*(*E*) measurements showed a high loss in capacitance values due to conductivity of lithium ions (Figure S11).

Temperature-dependent dielectric permittivity values were measured in the frequency ranges of $10^3 - 10^6$ Hz in the temperature range of 100-300 K and are plotted as dielectric constant ε' vs T as well as $1/\varepsilon'$ vs T in Figure 5a,b. An increase in the measured relative permittivity values was observed at all frequencies upon decreasing the temperature to 200 K. The dielectric anomaly correlated well with antiferroelectric behavior that emerges starting at 200 K as antiferroelectric materials are known to have large dielectric constants (Figure 5a). A plot of inverse dielectric permittivity vs T $(1/\varepsilon' \text{ vs } T)$ indicated that it follows the Curie–Weiss law, $\varepsilon' = C/(T - \theta)$ for T > 206 K, where C is a Curie constant and θ is a Curie– Weiss temperature. Based on the Curie–Weiss fit, $C = 1.1 \times$ 10^4 K and $\theta = -0.33$ K. The extremely low θ value was attributed to a series addition of capacitance values from both polar and nonpolar domains possibly due to the presence of grain boundaries thus making the resulting capacitance values much lower.

3.3. Origin of Ferroelectricity/Antiferroelectricity. The polar-to-antipolar phase transition shown upon cooling down to 90 K is the result of a change in the OOR direction across the interlayers, and the two phases compete, thus resulting in a diffuse phase transition in the SXRD data. In this system, both the SOJT mechanism, where d(Nb)-p(O) charge transfer induces a polarization in the NbO₂ layer, and the trilinear

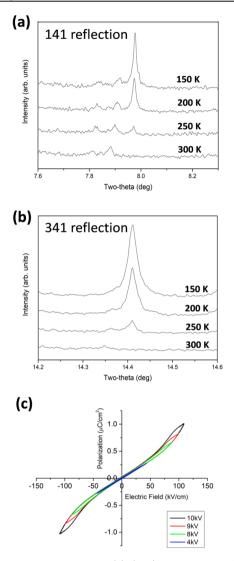


Figure 4. Thermal evolution of (a) (141) reflection and (b) (341) reflection as a function of temperature. (c) Polarization–electric field hysteresis measurement of $Li_2SrNb_2O_7$ at 77 K.

coupling of OOR and OOT distortions with A-site cation displacement played an important role in stabilizing the polar and antipolar phases. From Figure 1, the calculated layerresolved polarization values indicated that, for both A21am and Pnam phases, the polarization in the NbO₂ layers enhances the polarization in the SrO layers. This is in contrast to Sr₃Zr₂O₇, where polarization is merely contributed by A-site cation displacements induced by OOR and OOT distortions.¹³ Considering each distortion mode individually, calculations revealed that the out-of-phase OOT mode resulted in a net energy gain but the in-phase OOR mode did not result in energy minimization (Figure S12a). This observation is in sharp contrast to previous studies on hybrid improper ferroelectrics, where each of the modes resulted in significant energy gains. Calculations were also performed to determine the energetics of the coupling of OOR and OOT modes with the displacement of A-site cations (Figure S12b,c). Both polar and antipolar phases were found to be energetically most stable when they had a finite amplitude of OOR and OOT modes coupled with A-site displacements. This observation was corroborated with calculations on the energetic contributions of the different distortion modes which revealed that the

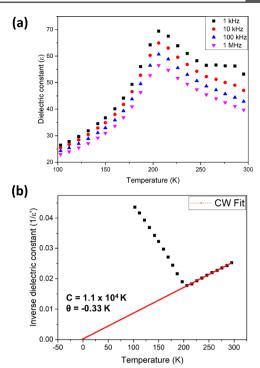


Figure 5. Temperature-dependent dielectric permittivity measurements of $\text{Li}_2\text{SrNb}_2\text{O}_7$ plotted as (a) ε' vs *T* and (b) inverse dielectric permittivity vs *T* at 1 MHz.

trilinear coupling of X_2^+ and X_3^- modes with the polar (or antipolar) modes resulted in the lowest energy structure (Figure S13). Hence, the hybrid improper ferroelectricity/ antiferroelectricity mechanism plays an equally important role in stabilizing the polar phase for Li₂SrNb₂O₇.

4. CONCLUSION

In summary, this study has revealed a new polar layered oxide, Li₂SrNb₂O₇, with an A2₁am structure at room temperature that shows a phase transition to an antipolar Pnam phase upon cooling to 200 K. On heating to high temperatures, it undergoes a phase transition to an unknown centrosymmetric phase at 550 K and a tetragonal I4/mmm structure at 650 K. Figure 6 shows the phase diagram of Li₂SrNb₂O₇ as a function of temperature. Temperature-dependent second harmonic generation revealed a noncentrosymmetric-to-centrosymmetric phase transition upon cooling from room temperature to 90 K. The polar-to-antipolar phase transition was also confirmed by polarization-hysteresis loop measurements which showed double hysteresis behavior at 90 K and an increase in the dielectric susceptibility at 200 K. The phase transition was attributed to a competition between hybrid improper ferroelectric/antiferroelectric mechanisms, where the polar/antipolar phases were stabilized by a trilinear coupling with oxygen octahedral distortions. Additionally, the SOJT mechanism also plays an important role in stabilizing the polar/antipolar phases in this system. The polar-to-antipolar phase transition offers a novel path to realize antiferroelectric materials in compounds that have energetically similar polar and antipolar structural phases. New materials discovered by this method can be interesting for applications in energy storage devices.

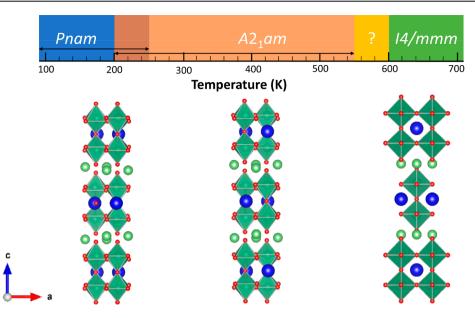


Figure 6. Phase diagram of Li₂SrNb₂O₇ as a function of temperature.

ASSOCIATED CONTENT

S Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.chemma-ter.9b00786.

Family tree diagram of all structures from first-principles calculations, energy-band structures of $Li_2SrNb_2O_7$, low temperature SHG measurement, refinements of SXRD and NPD data, TEM images showing stacking faults, additional results from electrical measurements, and results from first-principles calculations (PDF)

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Notes

The authors declare no competing financial interest.

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